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Prof. Anna Fedorovna KUZINA

100th Anniversary of birthday



K. E. G E R M A N
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ELECTROCHEMISTRY

100th Anniversary of birthday 1918 - 2018



In IPC AN USSR, 1956 – 1992 (after coming back from Ozersk)

Some fundamental chemistry of Tc Initial studies of technetium

irradiation of Mo in reactors for Tc (Krasnoyarsk) synthesis of new Tc compounds electrochemistry of technetium analytical chemistry of technetium

Separation of Tc at PA Mayak Conversion to Tc metal Cluster compounds of technetium

First micro-grams of Tc-99

Mo-98 (n, γ)Mo99 --- (β -decay)Tc-99 in the nuclear reactor Missions of Anna Kuzina to Krasnoyarsk Work with Anatoly Tsarenko



Laboratory of radiochemistry. Sitting: Albina Oblova, Anna Kuzina, academician Viktor Spisyn, Lyubov Barsova, Vitaly Kabanov, L. Troitsky, sdanding 1st line: Sergey Kryutchkov, Konstantin German, Valeria Pershina, A. Kisileva, Nina Budantseva, < Tamara Yurik, 2d line Sergey Kabakchi, 3rd line V. Mironov, Alexander Maslennikov, Al Vikhalin A. Kudryavtsev, R. Alimov, 1982, Moscow IPCAN USSR (now IPCE RAS)

First motivation for exploring Tc chemistry for the Closed Fuel Cycle

Tc-99 is a key dose contributor at HLW repositories if TRU elements are greatly reduced by recycling

- long half-life of Tc ($t_{1/2} = 2.14 \times 10^5$ years),
- high mobility, and solubility under oxidizing conditions

Methods for managing the long-term threat of Tc to the environment

- Stable waste form/repository system providing with strict limits for Tc release over a long period of time (~1 million years?).
- Transmutation of radioactive Tc to stable Ru im nuclear rectors.

Main problems of Tc

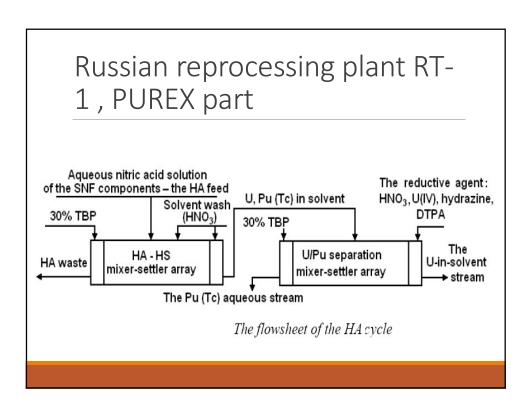
Tc is important item in Nuclear Industry

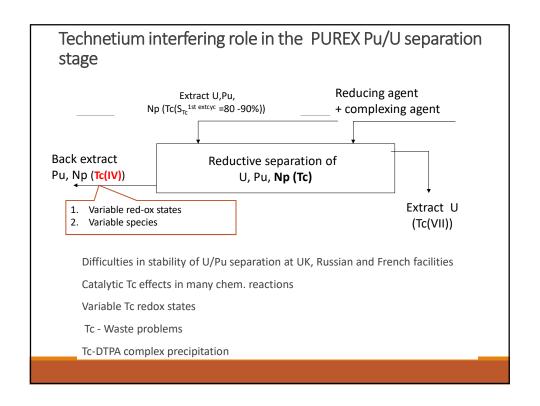
Tc redistribution in PUREX produces flows with long-lived high radioactive wastes

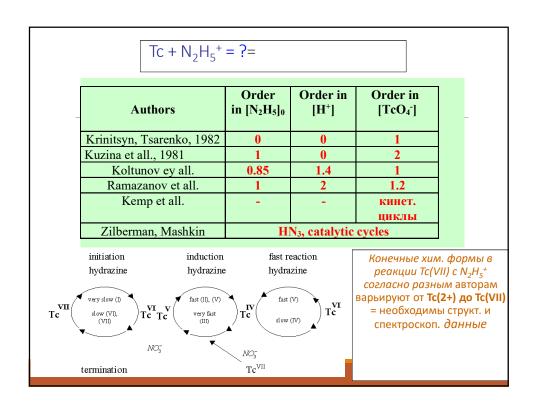
Tc interferes at U/Pu separation stage in PUREX process

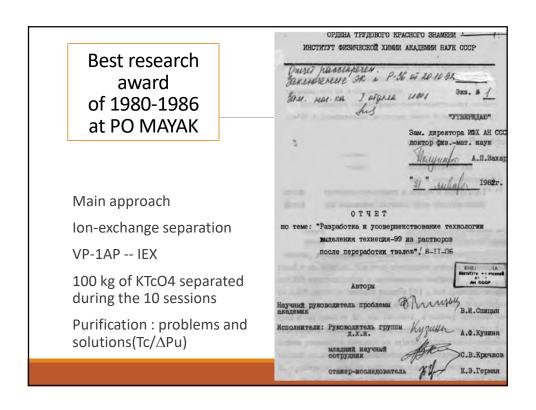
Tc accumulation in High burn-up fuel together with Mo, Ru, Rh

Tc in nuclear waste vitrification: Tc-Mo-Ru metal phases, Tc(VII) volatility









Development of ion-exchange technology for Tc separation in IPCE RAS (1971-1976)



Prof. A.F. Kuzina

(Tc Group leader till 1987)

presenting her Tc samples prepared in the Institute from the concentrate separated from radioactive wastes generated at Krasnoyarsk Reprocessing Plant to Glean SEABORG (1978)

Separation of macro amounts of Tc-99g in USSR





Prof. Anna KUZINA and acad. Victor SPITSYN analyzing the sample of Tc metal

- √1 kg of Tc was converted to metal in hot cell of IPCE RAS and distributed among different Russian institutes
- ✓In 1971-1976 IPC RAS in collaboration with Krasnoyarsk Mining Enterprise has separated from HAW some kilograms of K⁹⁹TcO₄
- ✓In 1983 -1986 collaboration of PO "Mayak", IPCE RAS and Radium Institute resulted in elaboration of anion-exchange technology for Tc separation and 40 kg of K⁹⁹TcO₄. This work was awarded with the special Diploma of the Russian authorities

Searching technetium applications

Anti-corrosion protection

Anti-fouling protection

Light-matter defectoscopy

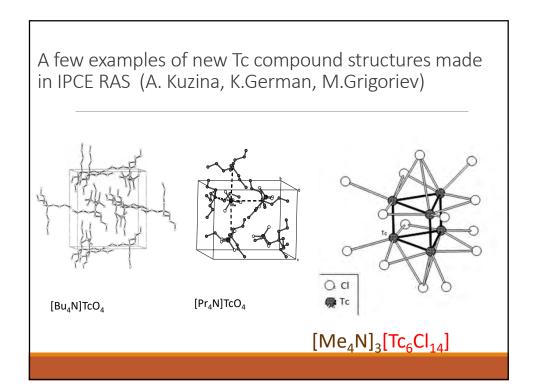
Catalysts

Ophtamo

Reference electro-current sources

Batteries

•••



The last work, ESCA, 1990, Italy, Tc&Re symp. (one of the most cited world publications on Tc)

X-ray photoelectron study of structure of technetium compounds

V.N. Gerasimov, S.V. Kryutchkov, K.E. German, V.M. Kulakov, A.F. Kuzina

«Technetium and Rhenium in Chemistry and Nuclear Medicine 3»

Technetium is rich in producing a great number of compounds. In synthesized chemical compounds it manifests a broad variety of valencies from -1 to +7. Therefore, various modern analytical methods are required to study the chemistry of technetium.

The present work reports the results of a study of technetium and its compounds using X-ray photoelectron spectroscopy (XPS), this being one of the most effective physical-chemical methods allowing us to solve the problems involved in investigating the electron structure of compounds, their chemical com-

compounds. Among the cluster compounds of d-transition elements the Tc clusters occupy a particular position. 46 On the one hand, this is due to their anomalous properties? and, on the other hand, to detection of new cluster structures for Tc, which reveal unusual systems of M-M bonds. 420 Clusters with both strong and weak field ligands are characteristic of technetium. In one study, 21 six basic structural modifications of Tc clusters with the weak field ligands were identified. In the present work these clusters were studied using the XPS method.

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- Anna Kuzina greatly supported international collaboration of IPC AN
- With Czechoslivakia,
- France, Poland, USA etc.
- A message from Roald Hoffman, 2018 :

Message to International Symposium on Technetium and Rhenium - Science and Utilization, 2018

The chemistry of rhenium and technetium is remarkably diverse and wide-ranging, given that one of the elements is among the rarest in the crust of the earth, and the other is almost entirely synthetic, the result of nuclear fission (I hope you like as much as I do, the image of nuclear physicists as synthetic chemists!) When these elements were first described nearly a hundred years ago, at much the same time, one could not have imagined that they would be used as superalloys and catalysts, tracers in medicine, or — in the Tc halide clusters — as examples of most unusual chemical bonding.

But they are so used, of value to our economy and well-being, a stimulus to our thinking. I wish the explorers of the Tc and Re worlds good science and great fun!

Roald Hoffmann, chemist and writer, who has worked, even if not much, with both Tc and Re chemistry, 08.08.2018 .

Walter Noddack, 125th Anniversary of birthday, 1893-2018



Discovery of RHENIUM

- + Walter Noddack
 - + Ida Tacke
 - + Otto Berg



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